

REGIOCONTROL IN THE SYNTHESIS OF OPTICALLY ACTIVE AMINO-4-PENTENEDIOLS
VIA EPOXY-4-PENTENOLS. NOVEL ACYCLIC ADENOSINE ANALOGUES¹

Walter Hümmer, Tibor Gracza, and Volker Jäger*

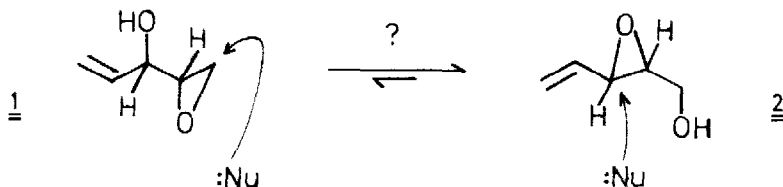
Institut für Organische Chemie der Universität
Am Hubland, D-8700 Würzburg

Summary: Regioisomeric erythro-1- and -3-amino-4-pentenediols are available from 1,2-epoxy-4-pentenol **1** by aminolysis of **1** or of the rearranged 2,3-epoxypentenol **2**. With adenine, D- and L-enantiomers of the nucleoside analogue **10** are obtained via the chloride **9** and the mesylate **12**, resp.

The optically active epoxypentenol **1**, introduced in 1985/6,^{2,3} has proven a versatile building block.²⁻⁵ Starting from divinylcarbinol (DVC), this secondary epoxy alcohol can be obtained in both enantiomeric forms and high purity^{2,4,6} - an inherent feature of the powerful Asymmetric Sharpless Epoxidation (A.S.E.).⁷ In addition, **1** (erythro) with dilute alkali undergoes "epoxide migration" (Payne rearrangement⁸);^{2,3,5} in the equilibrium mixture the regioisomer **2** (threo) predominates⁵ (1/2 3:97) which makes **2** an equally attractive, complementary building block.^{5,9}

Hydrolysis in slightly acidic medium, through attack at C-1 in **1** and at C-3 in **2**, respectively, has furnished opposite enantiomers of the corresponding pentenediols^{2,5,10} (stereocontrol at two levels²). Further, highly regioselective reductions of **1** and **2** have led to six out of eight 4-pentenediol isomers.⁵

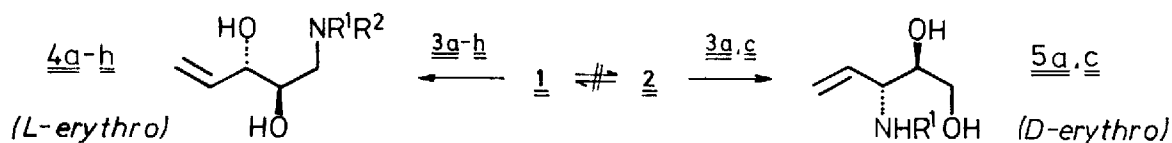
It is tempting to extend this scheme of regio/stereocontrol and apply other nucleophiles. This is promising if the basicity of these nucleophiles does not induce equilibration of the epoxides as stated in the hydrolysis of **1** by strong alkali,² or as it has proven crucial to achieve substitution at C-1 of 'normal' 2,3-epoxyalcohols.^{8b}



We now report that amines **3**¹²⁻¹⁵ specifically open the epoxide ring of **1** at the terminal site, and that of **2** at the allylic position, without interfering rearrangement! A selection of experiments is summarized in Table 1.

These results show that a broad range of amino functions - from ammonia, primary and secondary aliphatic amines, and aniline as a typical aromatic amine - can be added to **1** to furnish

Table 1. Aminopentenediols 4 and 5 from epoxy-4-pentenols 1 and 2, respectively.



Entry/ Epoxide	Amine 3	Conditions	Product/ Yield [%]	Physical Data (b.p. ^f [°C/Torr] or m.p. [°C]; [α] _D ²⁰)
1	a H ₃ N	a)	4a quant.	120-150/0.008; [α] _D ²² = -15.8 (c=2.217, MeOH)
2	b H ₂ NCH ₂ Ph	b)	4b 83	38; [α] _D ²⁰ = +8.8 (c=1.750, CHCl ₃)
3	c H ₂ NCHMePh (R)	b)	4c 96	40; [α] _D ²⁰ = +40.9 (c=1.370, CHCl ₃)
4	d H ₂ NCH(CH ₃) ₂	b)	4d 90	52-53; [α] _D ²² = -3.3 (c=0.675, CHCl ₃)
5	e H ₂ NC(CH ₃) ₃	b)	4e quant.	130-150/0.008; [α] _D ²² = -2.0 (c=2.425, CHCl ₃)
6	f HN(CH ₃) ₂	b)	4f 96 ^e	44 ^e ; [α] _D ²⁰ = -8.3 ^e (c=1.535, CHCl ₃)
7	g HN[CH(CH ₃) ₂] ₂	b)	4g 74	140-160/0.008; [α] _D ²² = +13.9 (c=0.905, CHCl ₃)
8	h H ₂ NPh	b)	4h 90	50-52; [α] _D ²² = +7.2 (c=1.245, MeOH)
9	a H ₃ N	c)	5a quant.	130-150/0.008; [α] _D ²² = +19.5 (c=0.435, MeOH)
10	c H ₂ NCHMePh (R)	d)	5c 90	120-130/0.005; [α] _D ²⁰ = +41.8 (c=0.9765, CHCl ₃)

a) With liquid ammonia, -78°C to room temp. over-night; similarly with conc. aqueous ammonia, room temp., 1 d. b) Room temp., 1 to 5 d; the epoxide 1 or 2 is dissolved in excess amine 3; work-up by fractional distillation. c) Conc. aqueous ammonia, room temp., 3 d. d) 100°C, 27 h. e) Yield of crude 4f, pure by NMR; data from material sublimed at 40°C/5 Torr, 52%. f) Kugelrohr.

1-amino-4-pentenediols 4. None of the regioisomers 5 expected from 2 could be detected in the crude products by NMR analysis. Likewise, when 2 was treated with ammonia 3a or phenethylamine 3c, the corresponding 3-aminodiols 5a,c were obtained in high yield and purity.¹⁶ The aminodiols 4 and 5 constitute vinyl-substituted aminopropanediols, and should be of pharmaceutical interest when compared to the many well-known aryloxypropanolamines (β-blockers).¹³

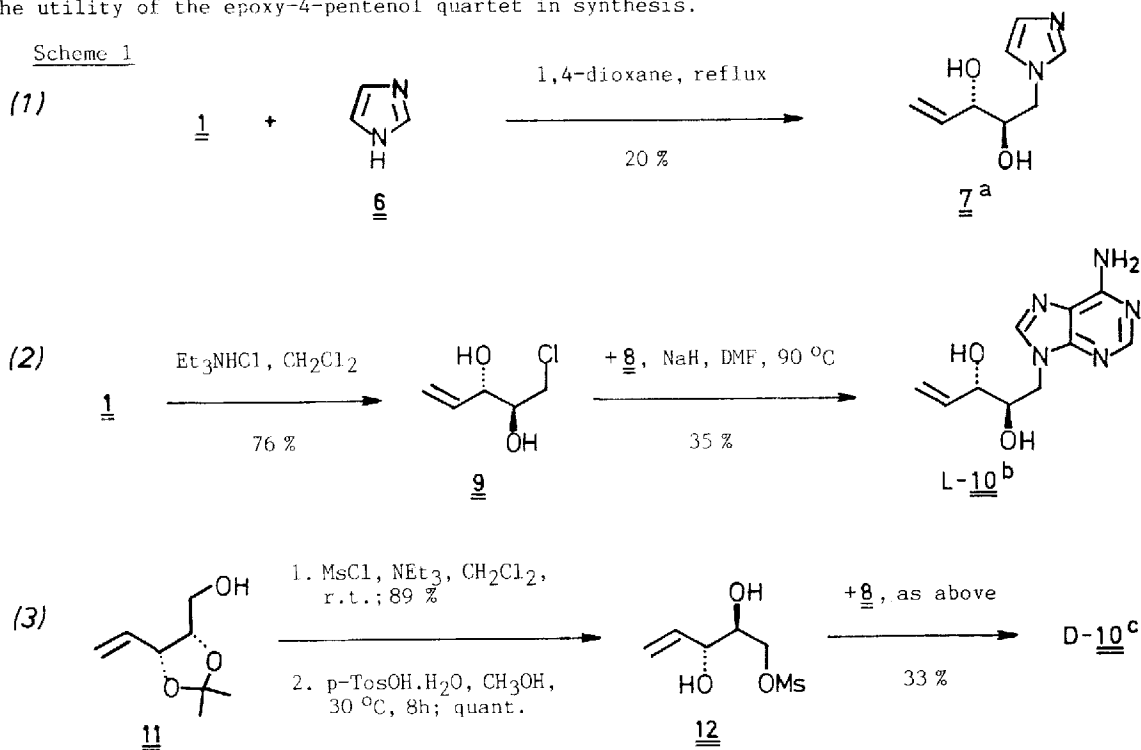
With respect to the enantiomeric purity of the new aminopentenediols, we note that this was not ascertained in most cases; we presume an e.r. of >96:4 for the 1-aminodiols 4, and of >93:7 for the 3-amino compounds 5 just as derived for the parent epoxides.^{2,6} We have sought to verify this using (R)-α-phenethylamine 3c (e.r. > 97:3); the third stereogenic centre thus added in 4c and 5c would serve as an 'e.r. indicator'. Indeed, with 4c and 5c, some of the ¹³C NMR signals showed a small neighbour peak of 3 to 6% intensity; these await identification (cp.¹⁶).

Next, the alkylation of less-reactive NH-heterocycles with 1 was explored, guided by results in the glycidol series.¹³ With imidazole 6, slow reaction was encountered which necessitated heating in 1,4-dioxane^{17a} to afford the diol 7 as an analytically pure oil in low yield, see Scheme 1, eq (1). Adenine (8) with sodium hydride in DMF at 100°C^{17a} did react with 1 but led to a brown mixture of at least 5 compounds. The latter problem could be solved by recurring to an earlier observation when amine hydrochlorides were tested to effect the above amination of 1: this had produced the (then undesired) 1-chloropentenediol 9. The sodium salt of

adenine 8,¹⁷ prepared as above, with 9 at 90°C in DMF was converted to a near quantitative mixture of N⁹-/N³-substituted products (ca. 3:1 by NMR). Crystallization of this solid from ethanol gave the analytically pure adenosine analogue L-10,¹⁸ see eq (2) in Scheme 1. The A.S.E. procedure would likewise be amenable to furnish the "natural" erythro compound D-10. In this case, however, we chose to rely on some of the ribonolactone chemistry we have developed recently,^{2,19} since this would secure maximum optical purity.⁶ Thus, the known triol acetonide 11^{2,19} was transformed to the mono-mesylate 12 and this, as above, added to the sodium salt of adenine 8. Again (vide supra), a mixture of alkylation products was formed which provided pure D-10 on recrystallization. The adenine derivatives L-10 and D-10 of "natural" and "unnatural" configuration, readily prepared in 100 mg-quantity, represent novel acyclic nucleoside analogues to be tested for antiviral activity.

In conclusion, the protocol of stereo- and regiocontrol outlined should broaden considerably the utility of the epoxy-4-pentenol quartet in synthesis.

Scheme 1



- a) 7: pale-yellow oil, spectroscopically pure after chromatography; $[\alpha]_D^{20} = +3.0$ ($c = 0.3535$, MeOH).
 b) L-10: colourless crystals, m.p. 216–218 °C (from ethanol); $[\alpha]_D^{20} = +2.8$ ($c = 0.375$, DMF).
 c) D-10: colourless crystals, m.p. 218–220 °C; $[\alpha]_D^{20} = -3.33$ ($c = 0.350$, DMF).

ACKNOWLEDGMENTS

This work was supported by the Fonds der Chemischen Industrie, Deutsche Forschungsgemeinschaft, and Bayer AG, Wuppertal. We are grateful to the A.v. Humboldt Foundation for awarding a post-doctoral fellowship to Dr. T. Gracza, and to BASF AG, Ludwigshafen, for chemicals.

REFERENCES AND NOTES

- Part of the (projected) Dissertation of W. Hümmer and post-doctoral work by T. Gracza 1987/8.
- V. Jäger, I. Müller, R. Schohe, M. Frey, R. Ehrler, B. Häfele, D. Schröter, Plenary Lecture at the 10th International Congress of Heterocyclic Chemistry, Waterloo, August 11-16, 1985; Lect. Heterocycl. Chem. 1985, 8, 79; B. Häfele, D. Schröter, V. Jäger, Angew. Chem. 1986, 98, 89; Angew. Chem. Int. Ed. Engl. 1986, 25, 87.
- S. Hatakeyama, K. Sakurai, S. Takano, J. Chem. Soc., Chem. Commun. 1985, 1759; Tetrahedron Lett. 1986, 27, 4485.
- R.E. Babine, Tetrahedron Lett. 1986, 27, 5791; S.L. Schreiber, T.S. Schreiber, D.B. Smith, J. Am. Chem. Soc. 1987, 109, 1529; K.C. Nicolaou, S.F. Webber, J. Ramphal, Y. Abe, Angew. Chem. 1987, 99, 1077; Angew. Chem. Int. Ed. Engl. 1987, 26, 1019; D. Askin, R.P. Volante, R.A. Reamer, K.M. Ryan, I. Shinkai, Tetrahedron Lett. 1988, 29, 277; D. Askin, R.P. Volante, K.M. Ryan, R.A. Reamer, I. Shinkai, Tetrahedron Lett. 1988, 29, 4245; S. Okamoto, Y. Kobayashi, H. Kato, K. Hori, T. Takahashi, J. Tsuji, F. Sato, J. Org. Chem. 1988, 53, 5590; J. Nokami, H. Ogawa, S. Miyamoto, T. Mandai, S. Wakabayashi, J. Tsuji, Tetrahedron Lett. 1988, 29, 5181.
- a) V. Jäger, R. Franz, W. Schwab, B. Häfele, D. Schröter, D. Schäfer, W. Hümmer, E. Guntrum, B. Seidel, Proc. IXth SCHC, Bratislava/CSSR, August 23-28, 1987, in 'Studies in Organic Chemistry', Vol. 35, "Chemistry of Heterocyclic Compounds", J. Kováč and P. Zálušký (Eds), pp 58-75, Elsevier, Amsterdam-Oxford-New York-Tokyo 1988. b) V. Jäger, D. Schröter, to be submitted.
- B. Koppenhoefer, M. Walser, D. Schröter, B. Häfele, V. Jäger, Tetrahedron 1987, 43, 2059; cp. ref. 2,3,4 (Schreiber; Askin).
- T. Katsuki, K.B. Sharpless, J. Am. Chem. Soc. 1980, 102, 5974; V.S. Martin, S.S. Woodard, T. Katsuki, Y. Yamada, M. Ikeda, K.B. Sharpless, J. Am. Chem. Soc. 1981, 103, 6237; J.G. Hill, K.B. Sharpless, Org. Synth. 1984, 63, 66.
- a) G.B. Payne, J. Org. Chem. 1962, 27, 3819 and references to earlier work. b) For reviews on applications see, f.e. H. Behrens, K.B. Sharpless, Aldrichim. Acta 1983, 16, 67; M.G. Finn, K.B. Sharpless in J.D. Morrison (Ed.), 'Asymmetric Synthesis', Vol. 5, 247, and E.E. Rossiter, *ibid.*, 193, Academic Press, New York 1985.
- S. Wershofen, H.-D. Scharf, Synthesis 1988, 854; cf. also: K.B. Sharpless, Janssen Chim. Acta 1988, 6, 3.
- All four 4-pentenetriol stereoisomers may now be prepared from DVC, when Scharf's results⁹ with Trost's stereo-retentive vinyl-epoxide carboxylation¹¹ of 2 are included.
- B.M. Trost, S.R. Angle, J. Am. Chem. Soc. 1985, 107, 6123.
- Reactions with amines are well-precedented only in the case of glycidol.¹³ With other epoxyalcohols^{8,14,15} mostly NH₂ was introduced via azide, occasionally through imino-ester cyclization.¹⁵ Similarly to the latter, RNH attachment from intermediate epoxyalkyl urethanes has been employed more often.^{8,14} These methods exhibit moderate to very high regio-selectivity, however, all but one¹⁵ being void of control.
- Survey: A. Kleemann, R.M. Wagner, "Glycidol, Properties, Reactions, Applications", p. 84 ff, Hüthig, Heidelberg-Basel-New York, 1981.
- N. Minami, S.S. Ko, Y. Kishi, J. Am. Chem. Soc. 1982, 104, 1109; W.R. Roush, M.A. Adam, J. Org. Chem. 1985, 50, 3752; M.C. Caron, K.B. Sharpless, J. Org. Chem. 1985, 50, 1560; J.M. Klunder, S.Y. Koo, K.B. Sharpless, J. Org. Chem. 1986, 51, 3710.
- B. Bernet, A. Vasella, Tetrahedron Lett. 1983, 24, 5491; R. Herzog, T. Herzig, B. Bernet, A. Vasella, Helv. Chim. Acta 1986, 69, 368.
- Representative and characteristic ¹³C NMR shifts (50.3 MHz, CD₃OD), f.e. for 4a: δ = 44.1 (t, C-1); 75.8, 76.2 (2xd; C-2, C-3); for 5a: δ = 64.6 (t, C-1); 75.8 (d, C-2); 57.5 (d, C-3) [ppm]; some additional peaks may be due to the presence of the threo isomers (\approx 3%), cf. Ref. 6.
- a) N. Ueda, T. Kawabata, K. Takemoto, J. Heterocycl. Chem. 1971, 6, 827; b) C.-I.J. Wang, S.H. Stam, J.M. Salvino, Tetrahedron Lett. 1988, 29, 1107.
- ¹³C NMR data of L-10 (50.3 MHz, d₆-DMSO): δ = 46.5 (C-1'), 72.5 (C-2'), 73.7 (C-3'), 115.5 (C-5'), 119.0 (C-5), 139.2 (C-4'), 142.5 (C-8), 150.2 (C-4), 152.7 (C-2), 156.3 (C-6) [ppm]; assignment based on excellent agreement with data reported for similar N⁹-/N³-substituted adenines: P. Holý, M. Bělohradský, I. Stibor, J. Koudelka, D. Šaman, J. Hodačová, A. Holý, J. Závada, Coll. Czech. Chem. Commun. 1987, 52, 2969.
- V. Jäger, B. Häfele, Synthesis 1987, 801.

(Received in Germany 28 December 1988)